

3-(3-Hydroxyphenyl)-1-(4-methoxyphenyl)-
prop-2-en-1-oneS. Sathiya Moorthi,^a
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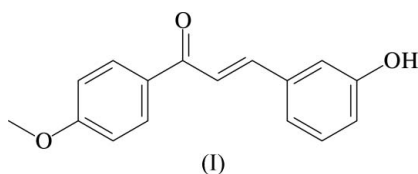
Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å
Disorder in main residue
 R factor = 0.055
 wR factor = 0.169
Data-to-parameter ratio = 12.6For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, $\text{C}_{16}\text{H}_{14}\text{O}_3$, crystallizes in the triclinic system, with two independent molecules in the asymmetric unit; these are related by a non-crystallographic twofold rotation axis. The two molecules differ slightly in the relative orientation of the two aromatic rings. They are linked by $\text{C}-\text{H} \cdots \pi$ interactions to form a dimer and $\text{O}-\text{H} \cdots \text{O}$ hydrogen bonds link the dimers into chains along the c axis. $\text{C}-\text{H} \cdots \pi$ interactions interlink the chains into a three-dimensional network.

Comment

Chalcones and their analogues are relatively easily available, not only by isolation from natural products but also by the methods of classical and combinatorial synthesis. Chalcones are being considered as anticancer agents, since they are natural compounds that are particularly cytotoxic towards K562 leukemia or melanoma cells (Sabzevari *et al.*, 2004). The cytotoxic, anticancer, chemopreventative and mutagenic properties of a number of chalcones have been reviewed by Dimmock *et al.* (1999). The antibacterial, fungistatic and fungicidal properties of these compounds have been reviewed by Opletalova (2000) and Opletalova & Sedivy (1999). Chalcones and their analogues are used as potential therapeutic agents in diseases of the cardiovascular system (Opletalova *et al.*, 2003). Photo-cross-linkable polymers having the chalcone moiety act as negative photoresist materials used in a wide variety of applications (Balaji *et al.*, 2003). Chalcones are also used in designing effective second-order non-linear optical materials (Zhao *et al.*, 2000). The structure determination of the title compound, (I), was undertaken as part of our study of chalcones.

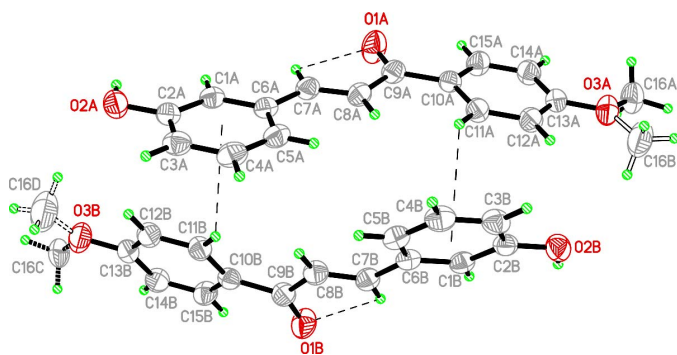


The asymmetric unit of (I) consists of two independent molecules which are related by a non-crystallographic twofold rotation axis. A displacement ellipsoid plot of the two independent molecules, *A* and *B*, is shown in Fig. 1. In both molecules, the methyl group is disordered over two different orientations, related by an approximately 180° rotation around the $\text{O}3-\text{C}13$ bond. The corresponding bond lengths and angles of the two molecules agree with each other and are comparable with those reported for the related structures 1-(4-chlorophenyl)-3-(4-hydroxyphenyl)prop-2-en-1-one (Ravi-

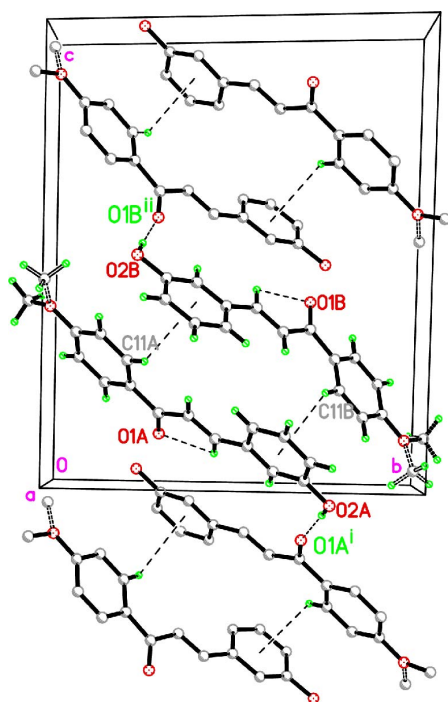
Received 7 February 2005

Accepted 14 February 2005

Online 26 February 2005


Figure 1

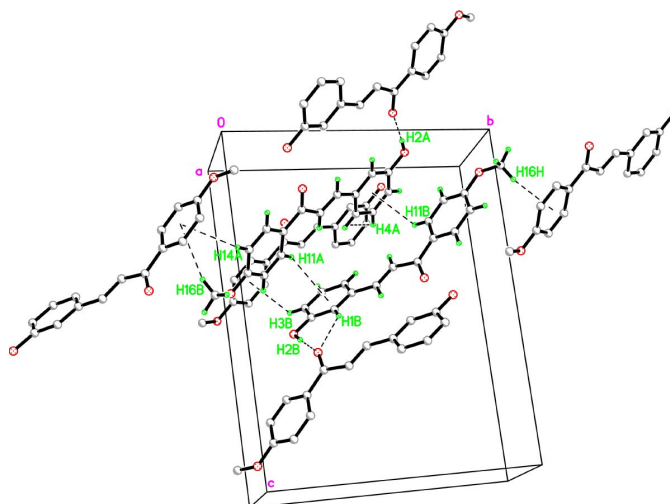
The structures of the two independent molecules of (I), showing the atomic numbering scheme and 30% probability displacement ellipsoids. Methyl groups are disordered with half-occupancy. Intramolecular hydrogen bonds and C—H... π interactions are shown as dashed lines.


Figure 2

A view of the C—H... π hydrogen-bonded dimers and O—H...O hydrogen-bonded chain formation. All the intra- and intermolecular interactions are shown as dashed lines.

shankar *et al.*, 2005) and 1-(4-hydroxyphenyl)-3-(4-methoxyphenyl)prop-2-en-1-one (Sathiya Moorthi *et al.*, 2005). The short H...H contacts which cause the narrowing or widening of neighbouring C—C—C angles from the normal value of 120° are H5A...H8A (2.27 Å), H5B...H8B (2.33 Å), H8A...H11A (2.11 Å) and H8B...H11B (2.14 Å). A significant conformational difference is observed between molecules *A* and *B*, in that the dihedral angle between the two aromatic rings is 6.3 (1)° in molecule *A*, whereas in molecule *B* the angle is 20.8 (1)°. The non-H atoms of the methoxy groups are almost coplanar with the attached rings.

In the molecular structure, each of the C7A—H7A...O1A and C7B—H7B...O1B interactions generates an *S*(5) ring motif. In the asymmetric unit, the *A/B* pair is linked by C—


Figure 3

Part of the crystal packing of (I), showing O—H...O, C—H...O and C—H... π interactions as dashed lines. Only the H atoms involved in the interactions are labelled.

H... π intermolecular interactions (C11A—H11A...Cg2 and C11B—H11B...Cg1, where Cg1 and Cg2 are the centroids of rings C1A—6A and C1B—6B, respectively), forming a dimer. In the crystal structure, the inversion-related dimers are linked by O2A—H2A...O1Aⁱ and O2B—H2B...O1Bⁱⁱ (symmetry codes are as given in Table 2) hydrogen bonds, forming a chain along the *c* axis (Fig. 2). The chain structure is further stabilized by weak C—H...O interactions (Table 2), whose geometry agrees with those reported by Jeffrey (1997) and Desiraju & Steiner (1999). Adjacent chains are interlinked by weak C—H... π interactions, forming a three-dimensional network (Table 2). Part of the network is shown in Fig. 3.

Experimental

The title compound was obtained by stirring a mixture of 4-methoxyacetophenone (4.15 g, 0.03 mol), 3-hydroxybenzaldehyde (3.66 g, 0.03 mol), a 20% solution of NaOH and ethanol at 273 K for 24 h, and neutralizing the mixture with dilute hydrochloric acid. The crude product was recrystallized from ethanol.

Crystal data

$C_{16}H_{14}O_3$	$Z = 4$
$M_r = 254.27$	$D_x = 1.286 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 5.6459 (9) \text{ \AA}$	Cell parameters from 3262 reflections
$b = 13.973 (2) \text{ \AA}$	$\theta = 2.4\text{--}26.7^\circ$
$c = 16.875 (3) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$\alpha = 88.517 (3)^\circ$	$T = 293 (2) \text{ K}$
$\beta = 85.574 (3)^\circ$	Block, colourless
$\gamma = 81.742 (3)^\circ$	$0.50 \times 0.41 \times 0.32 \text{ mm}$
$V = 1313.4 (4) \text{ \AA}^3$	

Data collection

Bruker SMART CCD area-detector diffractometer	4603 independent reflections
ω scans	3571 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.022$
$T_{\text{min}} = 0.957, T_{\text{max}} = 0.972$	$\theta_{\text{max}} = 25.0^\circ$
9588 measured reflections	$h = -6 \rightarrow 6$
	$k = -16 \rightarrow 16$
	$l = -18 \rightarrow 20$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.1135P)^2 + 0.018P]$
$R[F^2 > 2\sigma(F^2)] = 0.055$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.169$	$(\Delta/\sigma)_{\max} < 0.001$
$S = 1.04$	$\Delta\rho_{\max} = 0.24 \text{ e } \text{\AA}^{-3}$
4603 reflections	$\Delta\rho_{\min} = -0.19 \text{ e } \text{\AA}^{-3}$
365 parameters	
H-atom parameters constrained	

Table 1

Selected torsion angles ($^\circ$).

C16B—O3A—C13A—C12A	4.3 (5)	C16D—O3B—C13B—C12B	−3.8 (6)
C16A—O3A—C13A—C14A	−5.0 (4)	C16C—O3B—C13B—C14B	3.6 (4)

Table 2

Hydrogen-bonding geometry (\AA , $^\circ$).

Cg1, Cg2, Cg3 and Cg4 are the centroids of rings C1A–C6A, C1B–C6B, C10A–C15A and C10B–C15B, respectively.

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C7A—H7A \cdots O1A	0.93	2.40	2.758 (2)	102
C7B—H7B \cdots O1B	0.93	2.39	2.752 (2)	103
C11A—H11A \cdots Cg2	0.93	3.10	3.812 (2)	134
C11B—H11B \cdots Cg1	0.93	2.97	3.666 (2)	133
O2A—H2A \cdots O1A ⁱ	0.82	1.92	2.737 (2)	175
O2B—H2B \cdots O1B ⁱⁱ	0.82	1.93	2.747 (2)	176
C1B—H1B \cdots O1B ⁱⁱ	0.93	2.53	3.234 (2)	132
C16A—H16A \cdots O2B ⁱⁱⁱ	0.96	2.56	3.424 (5)	149
C3B—H3B \cdots Cg3 ^{iv}	0.93	2.96	3.644 (2)	131
C4A—H4A \cdots Cg4 ^{iv}	0.93	3.10	3.683 (2)	122
C14A—H14A \cdots Cg4 ^v	0.93	3.11	3.917 (2)	146
C16A—H16B \cdots Cg4 ^v	0.96	3.06	3.939 (5)	152
C16C—H16H \cdots Cg3 ^{vi}	0.96	2.88	3.697 (5)	144

Symmetry codes: (i) $1-x, 1-y, -z$; (ii) $1-x, 1-y, 1-z$; (iii) $2-x, -y, 1-z$; (iv) $1+x, y, z$; (v) $x, y-1, z$; (vi) $x-1, 1+y, z$.

In both molecules, the methyl group was found to be disordered over two orientations related by an approximate 180° rotation around the O3—C13 bond. The site occupancies of the disordered positions

C16A/C16B and C16C/C16D were initially refined to 0.547 (5)/0.453 (5) and 0.578 (8)/0.422 (5), respectively, and later fixed at 0.5 as the disordered atoms lie close to inversion centres. Corresponding bond lengths in the disordered components were restrained to be equal and the displacement parameters of the disordered atoms were approximated to isotropic behaviour. H atoms were placed in idealized positions and constrained to ride on their parent atoms, with O—H = 0.82 \AA , C—H = 0.93 or 0.96 \AA and $U_{\text{iso}}(\text{H}) = 1.2$ or 1.5 times U_{eq} of the carrier atom.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINTE* (Siemens, 1996); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

References

- Balaji, R., Grande, D. & Nanjundan, S. (2003). *React. Funct. Polym.* **56**, 45–57.
- Desiraju, G. R. & Steiner, T. (1999). *The Weak Hydrogen Bond in Structural Chemistry and Biology*. New York: Oxford University Press Inc.
- Dimmock, J. R., Elias, D. W., Beazely, M. A. & Kandepu, N. M. (1999). *Curr. Med. Chem.* **6**, 1125–1149.
- Jeffrey, G. A. (1997). *An Introduction to Hydrogen Bonding*. New York: Oxford University Press Inc.
- Opletalova, V. (2000). *Cesk. Farm.* **49**, 278–284.
- Opletalova, V., Jahodar, L., Jun D. & Opletal, L. (2003). *Cesk. Farm.* **52**, 12–19.
- Opletalova, V. & Sedivy, D. (1999). *Cesk. Farm.* **48**, 252–255.
- Ravishankar, T., Chinnakali, K., Nanjundan, S., Selvam, P., Fun, H.-K. & Yu, X.-L. (2005). *Acta Cryst.* **E61**, o405–o407.
- Sabzevari, O., Galati, G., Moridani, M. Y., Siraki, A. & O'Brien, P. J. (2004). *Chem. Biol. Interact.* **148**, 57–67.
- Sathya Moorthi, S., Chinnakali, K., Nanjundan, S., Radhika, R., Fun, H.-K. & Yu, X.-L. (2005). *Acta Cryst.* **E61**, o480–o482.
- Sheldrick, G. M. (1997). *SHELXTL*. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). *SMART* and *SAINTE*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Zhao, B., Lu, W.-Q., Zhou, Z.-H. & Wu, Y. (2000). *J. Mater. Chem.* **10**, 1513–1517.